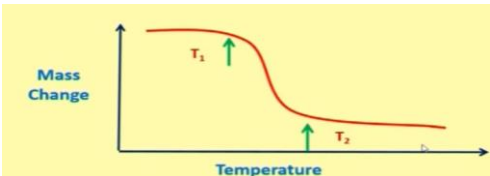


Name: Enrolment No:	
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UPES End Semester Examination, December 2023	
Course: Polymer Characterization & Testing Program: BSc by Research Course Code: CHEM 4018P	Semester: VII Time : 03 hrs. Max. Marks: 100
Instructions: <ol style="list-style-type: none"> 1. Write your enrolment number on the top left of the question paper. 2. Do not write any thing else on the question paper except your enrolment number. 3. Attempt all part of a question at one place only. 4. Internal choice is given for question number 9 of Section B and question number 11 of Section C only. 	

SECTION A (5Qx4M=20Marks)
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S. No.	Question	Marks	CO
Q 1	Explain, DSC, method to determine glass transition temperature.	4	CO1
Q2	How the tensile strength varies with the molecular weight of the polymer and how to choose appropriate molecular weight polymer from the graph?	4	CO2
Q3	Giving reason, arrange the following in increasing order of wavelength of U.V absorption. $\begin{matrix} \text{H}_2\text{C}=\text{C}-\text{C}=\text{CH}_2 \\ \quad \\ \text{H} \quad \text{H} \end{matrix} \quad \begin{matrix} \text{H}_2\text{C}=\text{C}-\text{C}=\text{C}-\text{C}=\text{C}-\text{CH}_2 \\ \quad \quad \quad \quad \\ \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \end{matrix} \quad \begin{matrix} \text{H}_2\text{C}=\text{C}-\text{C}=\text{C}-\text{C}=\text{CH}_2 \\ \quad \quad \quad \\ \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \end{matrix}$	4	CO1
Q4	What interpretation can be done from the given TGA Curve? How it is helpful? 	4	CO1
Q5	Discuss the Fire retardant property of polymer. PE blended with cellulose will be considered biodegradable or not? Justify your answer with suitable reason	4	CO3

SECTION B (4Qx10M= 40 Marks)

Q 6	Discuss the effect of elemental composition, molecular structure, crosslinking on COF? Why PE have low COF than PS?	10	CO2
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Q7	<p>A polymeric mixture is prepared by mixing three different polymers A, B, C having number average and weight average molecular weight and weight as given below</p> <p>Calculate the number average and weight average molecular weight of the Polymeric mixture.</p> <table border="1" data-bbox="240 449 1162 600"> <thead> <tr> <th>Polymer</th> <th>Mn</th> <th>Mw</th> <th>Wt. in mixture</th> </tr> </thead> <tbody> <tr> <td>A</td> <td>2.2×10^6</td> <td>4.8×10^6</td> <td>400</td> </tr> <tr> <td>B</td> <td>6.6×10^6</td> <td>8.6×10^6</td> <td>400</td> </tr> <tr> <td>C</td> <td>10×10^6</td> <td>10×10^6</td> <td>200</td> </tr> </tbody> </table>	Polymer	Mn	Mw	Wt. in mixture	A	2.2×10^6	4.8×10^6	400	B	6.6×10^6	8.6×10^6	400	C	10×10^6	10×10^6	200	10	CO2
Polymer	Mn	Mw	Wt. in mixture																
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C	10×10^6	10×10^6	200																
Q8	Discuss the biodegradation process in Polylactic acid (PLA).	10	CO3																
Q9	<p>What is your understanding about design for recycling? Give different ways by which chemical upcycling can be done.</p> <p>OR</p> <p>Discuss the different factors that affects the degradation of polymers .</p>	10	CO3																
SECTION-C (2Qx20M=40 Marks)																			
Q10	Explain the morphology of polymer with temperature variation mentioning glass transition temperature and the various other stages involved during transition. Also Discuss the importance of Tg and its relationship with Tm.	20	CO2																
Q11	<p>a) Taking example, discuss any two applications of U.V in Polymeric materials.</p> <p>b) How to calculate the degree of crystallinity in polymer sample? Also discuss the variation of toughness and density with degree of crystallinity.</p> <p>c) Comment on the type of peak obtained in crystalline, amorphous and semicrystalline polymers in XRD.</p> <p style="text-align: center;">OR</p> <p>a) Giving example, diagrammatically explain the different type of possible electronic transitions.</p> <p>b) Explain friction in polymers using two term model.</p> <p>c) Why in Polymer we have surface resistivity and volume resistivity? What is tracking resistance?</p>	6 8 6	CO1																