


Name:			
Enrolment No:			
UNIVERSITY OF PETROLEUM AND ENERGY STUDIES End Semester Examination, December 2022			
Course: Chemistry of Heterocyclic compounds Program: MSc. Chemistry Course Code: CHEM-8027P		Semester: III Time: 03 hrs. Max. Marks: 100	
Instructions:			
SECTION A (5Qx4M=20Marks)			
S.No.	Statement of question	Marks	CO
Q 1	Discuss the reactivity of Pyridine towards various electrophiles	4	CO1
Q2	Explain the difference in the dipole moment of pyrrole and pyrrolidine	4	CO1
Q3	Write a general mechanism for the synthesis of various heterocycles by Paal-Knorr synthesis	4	CO2
Q4	Pyridine ring is considered as π electron deficient. Explain with reasons	4	CO2
Q5	Explain with reasons which is more acidic among phenol and pyrrole?	4	CO3
SECTION B (4Qx10M= 40 Marks)			
S.No.	Statement of question	Marks	CO
Q1	Compare the stability of the following heterocycles on the basis of resonance a) Furan b) Thiophene c) Pyrrole d) Pyridine	2.5+2.5+2.5 +2.5	CO1
Q2	What is the oxidation state, and geometry of Iron in the heme complex of oxyhemoglobin and deoxyhemoglobin. Explain your answer based on MOT.	10	CO3
Q3	Write any one method for the synthesis of the following heterocycles a. 1,2,3-triazole b. Azepines	5+5	CO2
Q4	Compare the reactivity of Indole with nucleophile and electrophiles. Which is the most probable position of attack by these species?	10	CO3
OR			
	Discuss with mechanism the synthesis of Indole via Fischer Indole method and Reissert Method	5+5	CO3

SECTION-C
(2Qx20M=40 Marks)

S.No.	Statement of question	Marks	CO
Q1	a. Discuss the mechanism for the synthesis of Imidazole with α -hydroxy ketone and amidine. b. Explain the reactivity of pyridine with a) NaNH_2 and b) $n\text{-C}_4\text{H}_9\text{Li}$ c. Give reasons why thiophene is less reactive than benzene for electrophilic substitution? d. Explain with a valid reason why azepine is non-aromatic despite having a conjugated π -electron density?	5+5+5+5	CO2
Q2	a. Discuss the role of Sulfur and phosphorus ylides for the synthesis of various heterocycles. Explain the mechanism for the same. b. Elaborate on the porphyrin system present in hemoglobin? Draw the geometry of its complex with low spin and high spin Fe(II) .	10+10	CO3
OR			
	a. Discuss the biological relevance of Imidazole and Triazole rings b. Compare the rigidity of crown ether with porphyrin ring. How does it influence the metal coordination by these rings?	10+10	CO3