


Name:			
Enrolment No:			
UNIVERSITY OF PETROLEUM AND ENERGY STUDIES End Semester Examination, Dec 2022			
Course: Engineering Physics Program: B.Tech. CS (Batches 1-20) Course Code: PHYS 1023		Semester : I Time : 03 hrs. Max. Marks: 100	
Instructions: <ul style="list-style-type: none"> All questions are compulsory (Q. No. 9 and Q. No. 11 has an internal choice) All highlighted representations are vector quantities. Scientific calculators can be used for calculations. 			
SECTION A (5Q × 4M = 20 Marks)			
<ul style="list-style-type: none"> All questions are compulsory, Each Question carries 4 Marks Write very Short Answers/ Solve 			
Q. No.	Statement of question	Marks	CO
Q 1.	The surfaces $\rho = 3, \phi = 100^\circ, z = 3$ and $\rho = 5, \phi = 130^\circ, z = 4.5$ define a closed surface. Find the enclosed volume.	4	CO2
Q 2.	Outline Maxwell's equations in differential and integral forms for time variant fields.	4	CO2
Q 3.	Explain Ampere's Circuital law with proper diagram.	4	CO3
Q 4.	Calculate the de-Broglie wavelength of an α particle accelerated through a potential difference of 200 volts.	4	CO4
Q 5.	Explain quantum computing and its application.	4	CO5
SECTION B (4Q × 10M = 40 Marks)			
<ul style="list-style-type: none"> All questions are compulsory, Q 9. has an internal choice, Each Question carries 10 Marks Write Short/ Brief notes/ Derive/ Solve 			
Q 6.	Define Electric potential and establish a relation between electric potential and electric field intensity. Show that the Electrostatic field is a conservative field. (10)	10	CO2
Q 7.	(a) Explain Faraday's Law of induction. Apply Faraday's law to describe motional EMF. (4) (b) In a certain conducting region, $\mathbf{H} = yz(x^2 + y^2)\mathbf{a}_x - y^2xz\mathbf{a}_y + 4x^2y^2\mathbf{a}_z$ A/m. Determine the value of \mathbf{J} at $(5, 2, -3)$. (6)	10	CO3
Q 8.	(a) Mention any four differences between a classical computer and quantum computer. (4) (b) Given $ \psi\rangle = 3 0\rangle - 2i 1\rangle$. Find its normalized state. (6)	10	CO5
Q 9.	(a) What are the important conclusions that can be drawn from the Davisson and Germer's experiment? (5)		

	<p>(b) Calculate the lowest energy of an electron confined in a 3-D cubical box of each side 1 Å. (5)</p> <p style="text-align: center;">OR</p> <p>(a) Explain pair production. Why the pair production phenomenon cannot happen in empty space. (5)</p> <p>(b) The speed of a bullet of mass 50 g is measured to be 300 m/s with an accuracy of 0.01 %. With what accuracy can we locate the position of the bullet? (5)</p>	10	CO4
<p>SECTION-C (2Q × 20M = 40 Marks)</p> <p>• All questions are compulsory, Q.No. 11 has an internal choice, Each Question carries 20 Marks</p> <p>• Write long answer/ Derive/ Solve</p>			
Q 10.	<p>(a) Describe the construction and working of a gaseous laser system with proper diagram and label the components used. (10)</p> <p>(b) Describe the advantages and disadvantages of an optical fiber over conducting cables. A graded index fiber has a core diameter of 0.05 mm and numerical aperture of 0.02 at a wavelength of 8500 Å. Calculate the normalized frequency and the number of modes guided in the fiber. (10)</p>	20	CO1
Q 11.	<p>(a) Derive an expression for a normalized wave function for a particle of mass "m" moving in a one-dimension box of length L. Use schematic diagrams to analyze the behavior of wave function and probability density of wave function in the box. (10)</p> <p>(b) X-rays with $\lambda = 1\text{Å}$ are scattered from a carbon bock. The scattered radiation is viewed at 90° to the incident beam. Estimate the Compton shift. Calculate the kinetic energy imparted to the recoiled electron. (10)</p> <p style="text-align: center;">OR</p> <p>(a) Apply the general wave equation to derive an expression for the Schrodinger's time dependent equation in an operator form. (10)</p> <p>(b) A metallic surface, when illuminated with light of wavelength λ_1, emits electrons with energies upto a maximum value E_1, and when illuminated with light of wavelength λ_2, where $\lambda_2 < \lambda_1$, it emits electrons with energies upto a maximum value E_2. Prove that Planck's constant h and the work function ϕ of the metal are given by</p> $h = \frac{(E_2 - E_1)\lambda_1\lambda_2}{c(\lambda_1 - \lambda_2)} \quad \text{and} \quad \phi = \frac{E_2\lambda_2 - E_1\lambda_1}{(\lambda_1 - \lambda_2)} \quad (10)$	20	CO4

Constant	Standard Values
Planck's Constant (h)	6.63×10^{-34} Joule – sec
Permittivity of free space (ϵ_0)	8.85×10^{-12} Farad/meter
Velocity of light (c)	3×10^8 m/sec
Boltzmann constant (k_B)	1.38×10^{-23} JK ⁻¹
Rest mass of an Electron (m_o)	9.11×10^{-31} kg
Mass of the proton (m_p)	1.67×10^{-27} kg
Charge of an electron (e)	1.6×10^{-19} C