

| Q 7 | Apply Newton's forward interpolation to estimate the velocity at $x=0.4 \mathrm{~cm}$ for a fluid near a flat surface, given the velocity distribution provided below where $x$ represents the distance from the surface $(\mathrm{cm})$ and $v$ denotes the velocity $(\mathrm{cm} / \mathrm{s})$. |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Distance ( $x$ ) 0.1 | 0.3 | 0.5 | 0.7 | 0.9 |  |  |
|  | Velocity (v) 0.72 | 1.81 | 2.73 | 3.47 | 3.98 |  |  |
| Q 8 | The following system of equations is designed to determine concentrations (the $c$ 's in $\mathrm{g} / \mathrm{m}^{3}$ ) in a series of coupled reactors as a function of the amount of mass input to each reactor (the right-hand sides in $g / d a y$ ), $\begin{gathered} 15 c_{1}-3 c_{2}-c_{3}=3300 \\ -3 c_{1}+18 c_{2}-6 c_{3}=1200 \\ -4 c_{1}-c_{2}+12 c_{3}=2400 \end{gathered}$ <br> Execute two iterations of the Gauss-Seidel method with an initial approximation set as $\left[c_{1}, c_{2}, c_{3}\right]^{T}=[0,0,0]$. |  |  |  |  | 10 | CO3 |
| Q 9 | A ball at 1200 K is allowed to cool down in air at ambient temperature of 300 K . Assuming heat is lost only due to radiation, the differential equation for the temperature of the ball is given by $\frac{d \theta}{d t}=-2.2067 \times 10^{-12}\left(\theta^{4}-81 \times 10^{8}\right), \quad \theta(0)=1200 K$ <br> where $\theta$ is in $K$ and and $t$ in seconds. Determine the temperature at $t=240 \mathrm{~s}$ using the fourth order Runge-Kutta (RK) method, assuming a step size of $h=240 \mathrm{~s}$. <br> OR <br> Solve the boundary value problem $\left(1+x^{2}\right) y^{\prime \prime}+4 x y^{\prime}+2 y=2, y(0)=0, y(1)=$ $1 / 2$ by finite difference method. Use central difference approximations with $h=1 / 3$. |  |  |  |  | 10 | CO3 |
|  | $\begin{gathered} \text { SECTION C } \\ \text { (2QX20M=40 Marks) } \end{gathered}$ |  |  |  |  |  |  |
| Q 10 | The ideal gas law is given by $p v=R T$ <br> where $p$ is the pressure, $v$ is the specific volume, $R$ is the universal gas constant, and $T$ is the absolute temperature. This equation is only accurate for a limited range of pressure and temperature. Vander Waals came up with an equation that was accurate for larger range of pressure and temperature given by |  |  |  |  | 20 | CO1 |


|  | $\left(p+\frac{a}{v^{2}}\right)(v-b)=R T$ <br> where $a$ and $b$ are empirical constants dependent on a particular gas. Given the value of $R=0.08, a=3.592, b=0.04267, p=10$ and $T=300$ (assume all units are consistent), one is going to find the specific volume, $v$, for the above values. Without finding the solution from Vander Waals equation, what would be a good initial guess for $v$ ? Utilize Newton-Raphson method and conduct two iterations. Show all steps in calculating the estimated root, absolute relative approximate error for each iteration. |  |  |
| :---: | :---: | :---: | :---: |
| Q 11 | Solve the following Laplace equation $u_{x x}+u_{y y}=0$ numerically, using five-point formula and Liebmann iteration, for the following mesh with uniform spacing and with boundary conditions as shown below in the figure. Obtain the results correct to two decimal places. <br> OR <br> Solve by Crank-Nicolson method the following heat conduction equation $u_{t}=u_{x x}$ <br> subject to $u(x, 0)=0, u(0, t)=0$ and $u(1, t)=t$, for two time steps. | 20 | $\mathrm{CO4}$ |

