Name:

Enrolment No:



UNIVERSITY OF PETROLEUM AND ENERGY STUDIES

End Semester Examination, December 2022

Course: Industrial Chemicals and Environment Semester: V

Program: BSc Hons Chemistry

Course Code: CHEM3007D

Time : 03 hrs.

Max. Marks: 100

Instructions: Attempt all questions.

	SECTION A			
(5Qx4M=20Marks)				
S. No.		Marks	СО	
Q 1	At which temperatures (in degree Kelvin) will Nitrogen, Oxygen, Argon, and Helium separate in an ASU?	4	CO3	
Q 2	Write short notes on: A) Primary pollutants B) Secondary pollutants	4	CO2	
Q 3	Mention ways in which Bleaching powder and Hydrogen peroxide are used in the industry?	4	CO3	
Q 4	Discuss the importance of ultrapure metals in semiconducting industry, give examples.	4	CO3	
Q 5	How does a biocatalyst impact the activation energy of a reaction in comparison to a non-catalytic and a chemo-catalyst-based reactions?	4	CO1	
	SECTION B			
	(4Qx10M= 40 Marks)			
Q 1	Recall the different types of information given in the MSDS.	10	CO3	
Q 2	Enlist the different ways solid waste can be managed as part of their disposal process.	10	CO2	
Q 3	Differentiate, with diagram, between Induced Fit and Lock & Key mechanisms for Biocatalysis.	10	CO1	
Q 4	Elaborate on the different water pollutants, specially covering their sources and environmental impact.	10		
	Or		CO2	
	Discuss the following:			
	A) Techniques to measure air pollutantsB) Techniques to measure water pollutants	5 + 5		
	SECTION-C		•	

SECTION-C (2Qx20M=40 Marks)

Q 1	A) Elaborate the process of acid rain formation? Give chemical equations. Write a brief note on their environmental impact.B) Arrange biomedical wastes in various categories and assign the categories to appropriate color coding of waste bag.	10 + 10	CO2
	Or		
	A) Elaborate on the photochemistry of smog. Give equations.B) Elaborate on Effluent treatment plants and briefly touch upon its limitations.	10 + 10	
Q 2	 A) Distinguish between Gross Calorific Value and Net Calorific Value. In addition, mention how are they related to each other? B) Discuss the role of latent heat of condensation of steam in calculating Net Calorific Value of a fuel. C) Identify the different corrections applied to calculate Gross Calorific Value. D) Use the example of synthesis of Ibuprofen to demonstrate the difference between simple chemical reaction route and green synthesis inspired chemical reaction route. 	4+4+4 +8	CO1