Name:

**Enrolment No:** 



## UNIVERSITY OF PETROLEUM AND ENERGY STUDIES End Semester Examination, December 2022

Course: Polymer Chemistry Semester: III

Program: M.Sc Chemistry
Course Code: CHEM 8013
Time : 03 hrs.
Max. Marks: 100

**Instructions: Attempts all questions.** 

	SECTION A (5Qx4M=20Marks)				
S. No.	(DQAINI-ZUNZMINS)	Marks	СО		
Q 1	Define the following:				
	1. Compounding				
	2. Cold runner	4	CO1		
	3. Hot runner				
	4. Gate				
Q2	Polymer A is insoluble in water, discuss one polymerization techniques that you would suggest for its polymerization.	4 CO1			
Q3	Define composites. Discuss their classification on the basis of Matrix material	4	CO3		
Q4	Illustrate the importance of glass transition temperature.	4	CO3		
Q5	What is Internal Plastication? How it is better than addition of plasticizer?	4	CO3		
	SECTION B				
	(4Qx10M= 40 Marks)				
Q 6	To process PET into Bislery bottles, give the details of molding				
	technique.				
	Or	10	CO1		
	A wire has to be coated with Polymer. Discuss the details of the molding technique that	10	CO1		
Q7	Explain membrane Vapour Phase Osmometry, a method to determine molecular weight of a polymer.	10 CO2			
Q8	Discuss polyaddition polymerization reaction taking relevant example	10	CO2		
<b>Q</b> 9	Compare solubility characteristics of NaCl and Polyvinyl alcohol	10	CO2		

			SECTION-C (2Qx20M=40 Marks)		
Q 10	solution at 3 benzene have Conc (g/dl)  0.2716  0.194  0.1509  0.1235  0.1045  OR	Flow time (sec)  459.8  378.2  337.9  312.8  296.4	nd flow time data of PMMA sample in benzene the intrinsic viscosity of PMMA solution in	20	CO3
Q11	With the helmachine	5 15	CO1		