


Name:			
Enrolment No:			
UNIVERSITY OF PETROLEUM AND ENERGY STUDIES End Semester Examination, December 2022			
Course: Performance Analysis of Thermal Equipment Program: B Tech (RSEE) Course Code: EPEG 2017		Semester: III Time : 03 hrs. Max. Marks: 100	
Instructions:			
SECTION A (5Qx4M=20Marks)			
S. No.		Marks	CO
Q 1	What is a heat exchanger? Describe co-current, counter current and cross flow heat exchangers.	4	CO1
Q 2	Explain major modes of heat transfer in a boiler, furnace and a heat exchanger system.	4	CO1
Q 3	Find out the boiler efficiency and evaporation ration by direct method from the data given below. Type of boiler: Coal fired Quantity of steam (dry) generated: 8 TPH Steam pressure (gauge) / temp: 10 kg/cm ² (g)/ 180°C Quantity of coal consumed: 1.8 TPH Feed water temperature: 85°C GCV of coal: 3200 kCal/kg Enthalpy of steam at 10 kg/cm ² pressure: 665 kCal/kg (saturated) Enthalpy of feed water: 85 kCal/kg	4	CO2
Q 4	When it is beneficial to apply logarithmic mean temperature difference (LMTD) and when it is convenient to use effectiveness-NTU methods in the analysis of a heat exchanger.	4	CO1
Q 5	Describe the fluidization process in FBC boilers and their three main advantages over conventional firing boilers.	4	CO2
SECTION B (4Qx10M= 40 Marks)			
Q 6	Explain the features of a good steam distribution system.	10	CO2
Q 7	Explain the purpose of performance test of boilers. Define boiler efficiency and evaporation ratio and factors affecting these parameters during operations. Provide a detailed explanation of methods to evaluate boiler's efficiency with their merits and demerits.	10	CO2

Q 8	<p>Calculate the amount of heat required to convert 5.00 kg of ice at -10°C to steam at 100°C at atmospheric pressure. Given that:</p> <p>The specific heat capacity of ice = $2.10 \text{ kJ}/(\text{kg K})$ Latent heat of fusion of ice = $336 \text{ kJ}/\text{kg}$ The specific heat capacity of water = $4.20 \text{ kJ}/(\text{kg K})$ Latent heat of vaporization of water = $2250 \text{ kJ}/\text{kg}$</p> <p style="text-align: center;">OR</p> <p>Superheated steam at an average temperature 200°C is transported through a steel pipe ($k = 50 \text{ W}/\text{m}\cdot\text{K}$, $D_o = 8.0 \text{ cm}$, $D_i = 6.0 \text{ cm}$, and $L = 20.0 \text{ m}$). The pipe is insulated with a 4-cm thick layer of gypsum plaster ($k = 0.5 \text{ W}/\text{m}\cdot\text{K}$). The insulated pipe is placed horizontally inside a warehouse where the average air temperature is 10°C. The steam and the air heat transfer coefficients are estimated to be 800 and $200 \text{ W}/\text{m}^2\cdot\text{K}$, respectively. Draw equivalent thermal resistance diagram and calculate the daily rate of heat transfer from the superheated steam to the room.</p>	10	CO3
Q 9	<p>List major sources of heat loss and fuel economy/energy efficiency measures for an industrial furnace. (5 Marks)</p> <p>Find out the efficiency of reheating furnaces by direct method from the following data: Dimension of hearth of reheating furnace = $2\text{m} \times 4\text{m}$ Rate of heating of stock = $125 \text{ kg}/\text{m}^2 /\text{hr}$. Temperature of heated stock = 1030°C Ambient air temperature = 30°C Calorific value of fuel oil = $10200 \text{ kCal}/\text{kg}$ Specific gravity of fuel oil = 95 Fuel consumption during 8 hrs. of shift = 1980 liters. Mean specific heat of stock = $0.6 \text{ kCal}/\text{kg}/\text{K}$ (5 Marks)</p>	10	CO3
SECTION-C (2Qx20M=40 Marks)			
Q 10	<p>Describe various sources of waste heat, benefits of waste heat recovery, technologies available to recover waste heat and the end use of recovered heat. (10 Marks)</p> <p>Explain Cogeneration and Trigeration systems and their benefits over conventional power plants. (10 Marks)</p>	20	CO4
Q 11	<p>In a parallel flow heat exchanger, hot fluid enters the heat exchanger at a temperature of 150°C and a mass flow rate of $3 \text{ kg}/\text{s}$. The cooling medium enters the heat exchanger at a temperature of 30°C with a mass flow rate of $0.5 \text{ kg}/\text{s}$ and leaves at a temperature of 70°C. The specific heat capacities of the hot and cold fluids are $1.15 \text{ kJ}/\text{kg}\cdot\text{K}$ and $4.18 \text{ kJ}/\text{kg}\cdot\text{K}$, respectively. The convection heat transfer coefficient on the inner and</p>	20	CO4

outer side of the tube is $300 \text{ W/m}^2\cdot\text{K}$ and $800 \text{ W/m}^2\cdot\text{K}$, respectively. For a fouling factor of $0.0003 \text{ m}^2\cdot\text{K/W}$ on the tube side and $0.0001 \text{ m}^2\cdot\text{K/W}$ on the shell side, determine (a) the overall heat transfer coefficient, (b) the exit temperature of the hot fluid and (c) surface area of the heat exchanger.

OR

Hot water ($c_{ph} = 4.18 \text{ kJ/kg}\cdot\text{K}$) with mass flow rate of 2.5 kg/s at 100°C enters a thin-walled concentric tube counterflow heat exchanger with a surface area of 23 m^2 and an overall heat transfer coefficient of $1000 \text{ W/m}^2\cdot\text{K}$. Cold water ($c_{pc} = 4.18 \text{ kJ/kg}\cdot\text{K}$) with mass flow rate of 5 kg/s enters the heat exchanger at 20°C , determine (a) the heat transfer rate for the heat exchanger and (b) the outlet temperatures of the cold and hot fluids. After a period of operation, the overall heat transfer coefficient is reduced to $500 \text{ W/m}^2\cdot\text{K}$, determine (c) the fouling factor that caused the reduction in the overall heat transfer coefficient. Use *effectiveness-NTU* correlation provided below wherever required.

$$\text{Counter-flow} \quad \varepsilon = \frac{1 - \exp[-NTU(1 - c)]}{1 - c \exp[-NTU(1 - c)]} \quad (\text{for } c < 1)$$

$$\varepsilon = \frac{NTU}{1 + NTU} \quad (\text{for } c = 1)$$