Name:		14 LIDES			
Enrolment No:					
	UNIVERSITY OF PETROL	EUM AND ENERGY STUDIES			
	End Semester Examination, Dec 2021				
	Course: Chemistry of Heterocyclic Compounds Semester: III				
_	Program: M.Sc. Chemistry  Time 03 hrs.				
Course	Course Code: CHEM 8027P Max. Marks: 100				
	SEC	TION A			
	on A: Each Question will carry 4 Marks				
2. Instru	uction: Complete the statement / Select the c	orrect answer(s)			
S. No.	Quartiens			More	
S. NO.	Questions		CO	Max Marks	
Q 1	Discuss about the basicity of Pyridine		CO1	4	
Q 2	Which ring is more electrophilic among Benz	ene and Pyridine? Give reasons	CO1	4	
Q 3	Electrophilic aromatic substitution on 5 mem position. Give reasons	bered heterocycles is regioselective at the alpha	CO1	4	
Q4	Mention the reagents used for the conversion	of substituted amide to aryl aldehyde	CO1	4	
Q5	Which reagents are required for the nitration a substitution reaction.	and sulfonation of Furan by Electrophilic aromatic	CO1	4	
SECTION B					
1. Each	question will carry 10 marks				
2. Instru	uction: Write short / brief notes				

	Explain the mechanism of a reaction for the formation of amino alkylated 5-membered heterocyclic compounds.		10	
-	Discuss the chemical properties of Pyridine in terms of electrophilic and nucleophilic reactions		10	
-	ve an account of the nucleophilic substitution reactions of α-halogenated 6-membered erocycles	CO2	10	
Q 9 Iden	Thirty the products for the following reactions  NH <sub>3</sub> , 200°  NH <sub>3</sub> , 200°  NH <sub>3</sub> , 200°  NH <sub>3</sub> , 200°  OR  Implete the following reaction with a detailed mechanism  SO <sub>3</sub> /H <sub>2</sub> SO <sub>4</sub> Br  SO <sub>3</sub> /H <sub>2</sub> SO <sub>4</sub> Pr  Response to the following reactions with a detailed mechanism	CO2	10	
SECTION-C				
1. Each Question carries 20 Marks.				
2. Instruction: Write long answer.				

Q 10	Write the detailed mechanism of a Mannich Reaction.		20
	OR	CO3	
	Write the detailed mechanism of Vilsmeier-Haack Reaction		
Q 11	Explain the reason for the following conversion		20
	E ⊕ E E	CO3	