Name: **Enrolment No:** UNIVERSITY OF PETROLEUM AND ENERGY STUDIES End Semester Examination, May 2021 **Programme Name: B.Tech ADE** Semester : 8th Course Name : Finite element method Time :03 hrs **Course Code** : MECH4007P Max. Marks: 100 Nos. of page(s) :03 **SECTION A** S. No. Marks CO 01 Explain the difference between finite element method and classical methods. 5 **CO1** Explain the Rayleigh Ritz method. Q 2 5 **CO1** Why are the polynomials preferred as shape functions? Q 3 5 **CO1** O 4 Explain plane stress and plane strain problems. 5 **CO1** Explain the properties of the global stiffness matrix. Q 5 5 **CO1** Explain the terms nodes, primary nodes, secondary nodes and internal nodes. Q 6 5 **CO1 SECTION B** Q 7 A composite rod subjected to compression is modeled by two bar elements, as shown in Figure. Determine the nodal displacements and the axial stress in each element. $E_{\rm st} = 200 \, {\rm GPa}$ **CO2** 10 $A_{\rm st} = 4 \times 10^{-4} \ \rm m^2$ Steel Aluminum 3 20 kN $E_{al} = 70 \text{ GPa}$ $A_{al} = 2 \times 10^{-4} \text{ m}^2$ 1 m 1 m Q 8 A plane truss is loaded and supported as shown in Figure. Determine the displacements at the free end using finite element method. Take, E = 200 GPa and A $= 200 \text{ mm}^2$ **CO2** 10 3 m (1) 10 kN -20 kN

Q 9	For the beam shown in Figure, determine the nodal values at the point of applied load. Take $E = 200$ GPa. 500 N 10 cm 30 cm 1 cm 2 cm	10	CO3
Q 10	For the frame shown in Figure, determine the global stiffness matrix and load vector and apply the boundary conditions using the elimination approach. Take $E = 200$ GPa and Area = 2×2 cm ² 10 kN 4 m 4 m 5 m	10	CO3
Q 11	Determine the determinant of the Jacobian for the triangular element shown in Figure. $ \begin{array}{c} y \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ $	10	CO3

