Name:			14 LIDES					
			UPES					
Enrolment No:			UNIVERSITY WITH A PURPOSE					
	UNIVERSITY OF PETROLEUM AND ENERGY STUDIES							
End Semester Examination (Online Mode), Dec 2020								
Course: Green Chemistry Program: M. Sc. Chemistry			Semester: III Time: 3 hr					
Course Code: CHEM 8002			Max. Marks: 100)				
		SECTIO	$ON - A \qquad 6 \times 5 = 30 \text{ Marks}$					
1. Each Question (A+B parts) will carry 5 Marks								
2. Ins Q 1			Select the correct answer(s) E principles among twelve principles	CO1				
Q 1								
	of Green Chemistry. These are,,							
	,,							
Q 2	Fill in the blanks with last (7-12) FIVE principles among twelve principles CO1							
	of Green Chemistry. These are,,							
	,		-,					
Q 3	A: Write any two best definitions of Green Chemistry.							
	B: The need for Gree	en Chemistry	.					
Q 4	Choose correct biocatalysts from the following list. CO1							
	A: RX/AlCl ₃ D: HNO ₃ +H ₂ SO ₄							
	B: nano silver with	polymer base	E: Zeolites					
	C: aq. HCl/ΔT		F: NaOH/ΔT					
Q 5	A. Green chemistry	mproves	of chemical manufacturers.	CO2				
	a) Competitiveness	b) Easines	ss of production					
	c) Services d) Chemicals							
	B. The green synthesis methods should have							
	a) Low efficiency b) High harmful products							
	c) Low energy requir	ow atom efficiency						
Q 6	Select the correct Green solvents from the below list. CO2							
	A: Toluene	D: Ether	G: ethyl lactate J: Acetone					
	B: Hexane	E: Water	H: cyclo pentyl methylether					
	C: Glycerol	F: Lactone	I: 2-methyltetrahydrofuran					

	SECTION – B $10 \times 5 = 50 \text{ Marks}$ ch question (A+B parts) will carry 10 marks truction: Write short / brief notes			
Q 7	A: What are the main goals of Green Chemistry.			
	B: How Green Chemistry advances towards a sustainable future.			
Q 8	Illustrate with examples (differences and similarities) Green			
	Chemistry with Environmental Chemistry.			
Q 9	A: How Green Chemistry is interdisciplinary in nature ?			
	B: Describe few Environmental Laws.			
Q10	Describe with minimum two national and one international industrial			
	case study on wealth from waste (with highlights of Green Chemistry).			
Q 11	A: Explain about emerging Green Technologies.			
	B: Illustrate few next generation Catalyst Design methods.			
	A: Discuss about the characterization techniques with results presented from the following FIGURES presented.			
	1.00 0.75 0.75 0.00 0.75 0.00 0.00 0.00 0			
	(a) (b) Figure (a):UV-Visible spectra of silver nano particles (AgNPs) synthesized with onion (O), tomato(T), Acacia catechu (C). Figure (b): FTIR spectra of AgNPs synthesized with anion, tomato, Acacia catechu.			
	B: Discuss the solvent free synthesis methods with examples.			
	OR			

A: Explain any THREE methods of greener approaches for nanoparticle	CO2
synthesis.	
B: Depict selective method for the oxidation of 5-hydroxymethylfurfural	CO1
(HMF) by using hydrogen peroxide (H ₂ O ₂) and activated-carbon-supported	
ruthenium (Ru/AC) as the catalyst.	