

**Enrolment No:** 



**Semester: VII** 

Time 03 hrs.

## UNIVERSITY OF PETROLEUM AND ENERGY STUDIES

**End Semester Examination, December 2019** 

**Course: Petrochemical Process Technology** 

Program: B. Tech. CE+RP

Course Code: CHEG432 Max. Marks: 100

Instructions: Answer all the questions of a section, in sequence. Write legibly.

## **SECTION A** $(4 \times 5 = 20 \text{ Marks})$

S. No.		Marks	CO
Q 1	What are first, second and third generation petrochemicals? Give examples.	5	CO1
Q 2	Differentiate the mechanism of free radical addition polymerization from ionic polymerization mechanism.	5	CO2
Q 3	Write short notes on the production of SBR.	5	CO4
Q 4	What is the significance of integration of petrochemical plant with refineries?	5	CO5
	<b>SECTION B (4 x 10 = 40 Marks)</b>		
Q 5	What is steam reforming? Discuss its kinetics and derive rate expression for steam reforming.	2+8	CO2
Q 6	Describe the production process of ammonia from syngas.	10	CO3
Q 7	Explain the manufacturing process of Acrylonitrile butadiene styrene (ABS) with the help of a neat diagram.	10	CO4
Q 8	Describe the production process of nitric acid, with the help of a neat flowsheet.  Or  Describe the Solvay process of manufacturing soda ash, in detail.	10	CO5
	SECTION-C $(2 \times 20 = 40 \text{ Marks})$		
Q 9	Explain the production process of any two of the following, with the help of a neat flowsheet.  a) Acetic acid (from Methanol) b) Adiponitrile c) Styrene	10+10	CO3
Q 10	<ul> <li>a) With the help of a neat flowsheet, describe the synthesis of polyethylene terephthalate (PET).</li> <li>b) What is Nylon – 6,6? Explain its production process with a neat flowsheet.</li> </ul>	10 2 + 8	CO4