Name: Enrolme	ent No:		
	UNIVERSITY OF PETROLEUM AND ENERGY ST	UDIES	
	End Semester Examination, December 2019		
Course		emester: III	
_	Program: Engineering ThermodynamicsTime 03 hrs.		
		ax. Marks: 100	
Instruc	tions: All the questions are compulsory. Steam table is allowed		
	SECTION A (Attempt all the questions)		
S. No.		Marks	СО
Q 1	A refrigeration cycle operating as shown in Fig has heat transfer Qout = net work of Wcycle is 844 kJ.		
	Determine the coefficient of performance for the cycle.		
	Hot body System $Q_{in}$ $W_{cycle} = Q_{out} - Q_{in}$ Cold body	5	CO1
Q 2	Simplify the general forms of the mass and energy rate balances to descriprocess of blowing up a balloon. List all of your modeling assumptions.	be the 5	CO1

Q 3	Explain why do frozen water pipes tend to burst?	5	CO3
Q 4	A heat pump receives energy by heat transfer from the outside air at 0°C and		
	discharges energy by heat transfer to a dwelling at 20°C. Is this in violation of the	5	CO1
	Clausius statement of the second law of thermodynamics? Explain.		
	SECTION B (Attempt all of the following questions)		
Q 5	Steam enters a converging–diverging nozzle operating at steady state with p <sub>1</sub> =40 bar,		
	T $_1$ = 400°C, and a velocity of 10 m/s. The steam flows through the nozzle with		
	negligible heat transfer and no significant change in potential energy. At the exit,	10	CO2
	p2 = 15 bar, and the velocity is 665 m/s. The mass flow rate is 2 kg/s. Determine the		
	exit area of the nozzle, in m <sup>2</sup> .		
Q 6	At the beginning of the compression process of an air-standard Diesel cycle operating		
	with a compression ratio of 18, the temperature is 300 K and the pressure is 0.1 MPa.		
	The cutoff ratio for the cycle is 2. Determine (a) the temperature and pressure at the	10	CO3
	end of each process of the cycle, (b) the thermal efficiency, (c) the mean effective		
	pressure, in MPa.		
Q 7	Water initially a saturated liquid at 100°C is contained within a piston–cylinder		
	assembly. The water undergoes a process to the corresponding saturated vapor state,		
	during which the piston moves freely in the cylinder. There is no heat transfer with		
	the surroundings. If the change of state is brought about by the action of a paddle		
	wheel, determine the net work per unit mass, in kJ/kg, and the amount of entropy		
	produced per unit mass, in kJ/kg K.	10	CO4
	OR		
	At the beginning of the compression process in an airstandard Otto cycle, $p1=1$ bar		
	and $T1=300$ K. The maximum cycle temperature is 2000 K. Plot the net work per		
	unit of mass, in kJ/kg, the thermal efficiency, and the mean effective pressure,		
	in bar, versus the compression ratio ranging from 2 to 14.		
Q 8	The absolute pressure inside a tank is 0.4 bar, and the surrounding atmospheric		
	pressure is 98 kPa. What reading would a Bourdon gage mounted in the tank wall	10	CO1
	give, in kPa? Is this a guage or vacuum reading?		
	SECTION-C(Attempt all of the following)		
Q 9	A vertical piston–cylinder assembly containing 0.05 kg of ammonia, initially a		
	saturated vapor, is placed on a hot plate. Due to the weight of the piston and the		
	surrounding atmospheric pressure, the pressure of the ammonia is 1.5 bars. Heating		
	occurs slowly, and the ammonia expands at constant pressure until the final	•	COL
	temperature is 25°C. Show the initial and final states on	20	CO4
	T-v and $p-v$ diagrams, and determine		
	(a) The volume occupied by the ammonia at each state, in m3.		
	(b) The work for the process, in kJ.		

A power plant based on the Rankine cycle is under development to provide a net power		
output of 10 MW. Solar collectors are to be used to generate Refrigerant 22 vapor at		
1.6 MPa, 50°C, for expansion through the turbine. Cooling water is available at 20°C.		
Specify the preliminary design of the cycle and estimate the thermal efficiency and the		
refrigerant and cooling water flow rates, in kg/h.		
OR		
Steam enters a turbine with a pressure of 30 bar, a temperature of 400°C, a velocity of 160 m/s. Steam exits as saturated vapor at 100°C with a velocity of 100 m/s. At steady	20	CO3
state, the turbine develops work at a rate of 540 kJ per kg of steam flowing through the turbine. Heat transfer between the turbine and its surroundings occurs at an average outer surface temperature of 350 K.		
Develop a full accounting of the net exergy carried in by the steam, per unit mass of		
steam flowing. Neglect the change in potential energy between inlet and exit. Let $T_0 = 25^{\circ}$ C, $P_0 = 1$ atm.		
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