UNIVERSITY OF PETROLEUM AND ENERGY STUDIES UPES

Program: B.Tech MSNT	Semester –	\mathbf{V}
Subject (Course): Material Characterization II	Max. Marks	: 100
Course Code : MTEG 342	Duration	: 3 Hrs
No. of page/s: 3		

Instructions:

- 1. All questions are compulsory.
- 2. This question paper has three sections; Section A, Section B, and Section C
- 3. In section A there are total 5 questions, each carrying 4 marks
- 4. In Section B there are total 4 questions, each carrying 10 marks. Question no. 9 is having an internal choice.
- 5. In Section C there are total 2 questions, each carrying 20 marks. Question no. 11 is having an internal choice.
- 6. Draw suitable diagrams wherever required.
- 7. Your answer should be concise and to the point.

Values of some physical constants:

Velocity of light, $c = 3 \times 10^8 m s^{-1}$
Mass of electron, $m_e = 9.1 \times 10^{-31} kg$;
Rydberg constant (R_H)=109678 cm ⁻¹ ;
Planck's Constant (h) = 6.6×10^{-34} Jsec;

Charge of electron, $e = 1.6 \times 10^{-19} C$ Mass of proton/neutron = $1.67 \times 10^{-27} kg$ Boltzmann Constant (K_B) = $1.38 \times 10^{-23} JK^{-1}$ 1 a.m.u. = $1.67 \times 10^{-27} kg$

	SECTION A (All questions are compulsory)						
Ques 1	With suitable diagram explain the ideal tip geometry requirement for achieving the maximum resolution in a scanning tunneling microscope.			[4]	CO1		
Ques 2	Discuss the step wise process of generating the topographic information of a surface when sample is scanned in an intermittent contact mode AFM.		[4]	CO2			
Ques 3	The different regions of electromagnetic spectrum (in Hz) are given below;						
	X-rays	Ultraviolet	Visible	Infrared (IR)	Microwave	[4]	CO3
	10 ¹⁸ to 10 ¹⁶	10 ¹⁶ to 10 ¹⁴	8×10^{14} to 3×10^{14}		10 ¹² to 10 ⁸	[.]	000
	Evaluate the respective spectroscopic wave number (cm ⁻¹) range.						
Ques 4	The normal vibrational modes of CO_2 have two degenerate bending modes each with a frequency corresponding to 526 cm ⁻¹ . A light of frequency corresponding to 1052 cm ⁻¹ is used to excite the ground state of CO_2 . According to harmonic oscillator model, what will be the outcome?		[4]	CO3			

Ques 5	How many number of rotational (R), translation (T) and vibrational (V) degrees of freedom are available for: $C \equiv C - (C \equiv C)_n - C \equiv CH$. (Hint: write V in terms of n)	[4]	CO4		
	SECTION B (All questions are compulsory)				
Ques 6	 (i) If at the top of STM tip, two atoms are at same distance from sample surface what artifacts will be observed in image. (ii) An electron with a total energy of E₀= 5 eV approaches a potential barrier with a height of 10 eV. If the width of the barrier is 0.18 nm, evaluate the percentage change in probability when barrier height is doubled. 		CO1		
Ques 7	Elucidate the tip requirement and working of following non-topography imaging mode: "Electric force microscopy (EFM)"		CO2		
Ques 8	 (i) If one wish to design absorption experiment (IR, UV-Visible, Microwave etc), what are the main components required along with their desired properties. (ii) A diffraction grating has a ruled area that is 10 cm wide and 500 lines/ cm. Determine the resolving power and resolution in the 6th and 9th order for 33333 cm⁻¹. 		CO3		
Ques 9	 (i) Elaborate on the analytical capabilities of X-ray photoelectron spectroscope (XPS) which makes it preferred tool for material characterization. (ii) Consider a C atom with ground state, first and second exited state binding energies of 290 eV, 20 eV and 10 eV respectively. Illustrate the XPS spectrum in both binding energy (BE) and kinetic energy (KE) scale assuming Al source (1486.6 eV). 				
	ORVacuum Cond. Band E_F The figure shows the energy level diagram (binding energies are indicated (1844, 154, 104, 18, 6, 4, 0 eV)) of a solid Si. If Si is bombarded with 4 keV X-rays. Discuss all possible outcomes of the process along with their energies (Hint: Photoelectrons, Auger electrons, X-rays etc.)Vacuum Cond. Band E_F Image: Cond. Band GoldValence BandImage: Cond. Band GoldImage: Cond. Band GoldImage: Cond. Band GoldValence LongImage: Cond. Band GoldImage: Cond. Band GoldImage: Cond. Band Gold <t< td=""><td>[10]</td><td colspan="3">CO4</td></t<>	[10]	CO4		

	SECTION C (All questions are compulsory)		
Ques 10	(i) The SPM's can be used to manipulate atoms to form desired structures at the atomic level. Explain the process of atomic patterning using:		CO1
	(a) Scanning tunneling Microscopy(b) Dip-Pen Lithography	[8+8+4]	CO2
	(ii) Calculate the broadening Δv of transitions in HCN at 25°C due to the Doppler effect in regions of the spectrum typical of rotational transitions (10 cm ⁻¹), vibrational transitions (1500 cm ⁻¹) and electronic transitions (60 000 cm ⁻¹).		CO3
Ques 11	 (a) Illustrate the normal modes of vibrations seen in a linear triatomic molecule using CO₂ as example. (b) Define selection rules and draw the microwave and Raman spectra for a diatomic molecule. Also discuss the origin of stokes and anti-stokes lines. (c) The separation between the first and second line in a microwave spectra of CO molecule (rigid rotor model) is 3.82 cm⁻¹. If same separation is obtained for H₂ molecule in Raman spectra. Evaluate the bond distance in both cases. 	[8+8+4]	CO4
	OR		
	 (a) Highlight the major difference in terms of interaction with electromagnetic radiations in Infrared (IR) and Raman Spectroscopy. Also comment why rotational modes of H₂ molecule are observed in Raman spectra and not in microwave spectra. (b) What are the major drawbacks of harmonic model for diatomic molecule? Using 	[10+10]	
	(b) what are the major drawbacks of harmonic model for diatomic molecule? Using an-harmonic (Morse) oscillator model explain different regions of Morse curve and draw the allowed energy levels and transitions.		