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Roll No. ....



## UNIVERSITY OF PETROLEUM AND ENERGY STUDIES

Mid Semester Examination - May, 2019

<b>Program/Course</b> : M. Tech Chemical (Spl. in Process Design)	Semester: II Max. Marks: 100
Subject: Process Modelling and Simulation	
Code: CHPD7009	
No. of pages: 2	<b>Duration</b> : 3 hrs
NOTE:	

## (A) OPEN BOOK and OPEN NOTES EXAMINATION

(B) Assume all missing data. State your assumptions clearly. Sketch wherever necessary.

## ANSWER ALL QUESTIONS

1. A stainless-steel rod (18% Cr, 8% Ni) 6.4 mm in diameter is initially at a uniform temperature of 25°Cand is suddenly immersed in a liquid at 150°Cwith  $h = 120 \text{ W/m}^2$ °C. Using the lumped-capacity method of analysis, calculate the time necessary for the rod temperature to reach 120°C. [20 marks]

Also, write a MATLAB code for solving the above problem with the data provided. [10 marks]

2. A tubular chemical reactor of length L and cross section of  $1 \text{ m}^2$  is employed to carry out a first order chemical reaction in which a material A is converted a product B with specific reaction rate constant  $k \text{ s}^{-1}$ . The feed rate is  $u \text{ m}^3/\text{s}$ , the feed concentration of A is  $c_0$ . The diffusivity of A is assumed to be constant at D m<sup>2</sup>/s. If there is a volume change during the reaction, develop a mathematical model in the form of a differential equaion for the concentration of A as a function of length along the reactor. [20 marks]

Also, write a MATLAB code for solving the above problem with the data provided. [10 marks]

3. Experimental setup of the batch vacuum distillation under total reflux condition in rotating packed bed (RPB) is shown in Fig. 1. The RPB was packed by wave thread stainless steel wire mesh, with a voidage of 0.95 and a specific surface area of 930  $\text{m}^2/\text{m}^3$ , respectively. The inner diameter, outer diameter and the axial height of the rotor were 180, 285 and 310 mm respectively. The rotor speed varied between 275 and 700 rpm, and thus providing 20 - 130 times gravitational force at the arithmetic mean radius.

The vapor entered axially into the RPB due to the pressure drop and left from the top of the rotor finally. The concentration of feeding ethanol was ranging from 70 to 95 wt%. The concentration of outlet ethanol was measured by gas chromatography (GC-S7900). The liquid flux was pumped from the effluent tank into the inner side of the rotor through a liquid distributor, and moved outward through the rotating packing under the centrifugal force. Thus, tiny droplets and thin films were formed at the surface of packing and resulted in large gas - liquid interfacial area, which was beneficial to the distillation. The vapor, containing high concentration of ethanol, was finally collected by condenser, and was then refluxed from the top of the RPB, while the water-rich liquid was discharged from the bottom and then returned to the reboiler. During the experiment, the operation pressure varied from 11.325 to 101.325 kPa and was controlled by a vacuum pump.

Derive a complete balance model equations starting from overall mass balance. Also suggest some numerical techniques to solve your model equations. [40 marks]

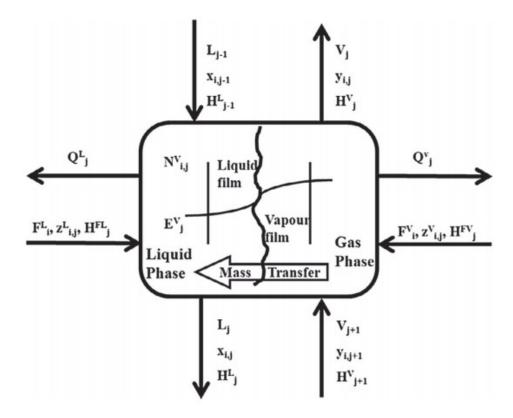


Figure 1: Non-equilibrium stage for vacuum distillation

## Nomenclature

a	gas-liquid interfacial area (m²/m³)	Greek le	tters
at	total specific surface area of packing $(m^2/m^3)$	3	packing voidage
ap	surface area of the 2 mm diameter bead per unit volume	ε <sub>L</sub>	liquid holdup
P	of the bead (1/m)	ρ	density (kg/m <sup>3</sup> )
Ср	heat capacity (m <sup>3</sup> /kg)	δ <sub>C</sub>	critical surface tension (N/m)
D <sub>G</sub>	diffusivity coefficient of gas (m <sup>2</sup> /s)	δω	surface tension of water (kg/s <sup>2</sup> )
DL	diffusivity coefficient of liquid $(m^2/s)$	ω	angular velocity(rad s <sup>-1</sup> )
dp	diameter of packing pore (m)		
E	heat transfer rate (J s <sup>-1</sup> )	Dimensi	onless groups
F	feed flow rate (mol $s^{-1}$ )	Fr <sub>L</sub>	liquid Froude number $(L^2a_t/g_c)$
h	heat transfer coefficient (w/m <sup>2</sup> /k)	Gr <sub>G</sub>	gas Grashof number $(dp^3g_c/v_G^2)$
HV	vapor enthalpy ( $J \text{ mol}^{-1}$ )	GrL	liquid Grashof number $(dp^3g_c/v_L^2)$
HL	liquid enthalpy $(J mol^{-1})$	Reg	gas Reynolds number $(G/a_t v_G)$
jн	the heat transfer factor	ReL	liquid Reynolds number $(L/a_t v_L)$
j <sub>D</sub>	the mass transfer factor	Sci	liquid Schmidt number $(v_L/D_L)$
K	phase equilibrium	WeL	liquid Webber number $(L^2 \rho_L / a_t \delta)$
k <sub>G</sub>	gas phase mass transfer coefficient (m/s)	Ка	the Kapitza number $(\mu^4 g/\sigma^3 \rho)$
k <sub>L</sub>	liquid phase mass transfer coefficient (m/s)	Nu	Nusselt number (hL/k)
L	liquid flow rate (mol $s^{-1}$ )	Pr	Prandtl number (µCp/k)
m	mass flow $(\text{kg s}^{-1})$	Sc	Schmidt number ( $\mu/\rho D$ )
N	mass transfer rate (mol $s^{-1}$ )	St	Stanton number (Nu/Re/Pr)
Q	heat exchange (w)	Sh	Sherwood number (k/L/D <sub>AB</sub> )
Q1	the power of condenser (w)	ψ	$c^{2}/(d+c)^{2}$
Ro	outer radius (m)		
Ri	inner radius (m)	Supersci	ript and Subscript
T	tempture (K)	i, m	component
u <sub>b</sub>	velocity (m/s)	j.,	the j plate
v	vapor flow rate (mol $s^{-1}$ )	L	liquid phase
Vi	volume inside the inner radius of the bed (m <sup>3</sup> )	v	vapor phase
vo	volume between the outer radius of the bed and the sta-	F	feed
0	tionary housing (m <sup>3</sup> )	e	equilibrium
vt	total volume of the RPB (m <sup>3</sup> )	N	the N plate
x	liquid mole fraction		the replace
У	vapor mole fraction		
z	feed mole fraction		
∆p	pressure drop (Pa) 2		