

Name:

Enrolment No:



## UNIVERSITY OF PETROLEUM AND ENERGY STUDIES

End Semester Examination, December 2018

Programme Name: B.Tech CE+RP

Course Name : Petroleum Refining Technology

Course Code : CHEG 437

Nos. of page(s) : 3

Semester : VII

Time : 03 hrs.

Max. Marks: 100

Instructions: (1) Assume suitable Data wherever necessary

### SECTION A (Attempt all Questions)

S. No.		Marks	CO																		
Q 1	Discuss about different tests and properties of aviation turbine fuel in brief.	10M	CO2																		
Q 2	Consider the following crude assay which has API =29 <table border="1" data-bbox="207 877 610 1350"><thead><tr><th>Volume %</th><th>TBP (°C)</th></tr></thead><tbody><tr><td>0</td><td>216</td></tr><tr><td>10</td><td>243</td></tr><tr><td>30</td><td>268</td></tr><tr><td>50</td><td>284</td></tr><tr><td>70</td><td>304</td></tr><tr><td>90</td><td>318</td></tr><tr><td>95</td><td>327</td></tr><tr><td>100</td><td>334</td></tr></tbody></table> <p>Plot the TBP data and calculate the volume average boiling point, mean average boiling point empirically and the characterization factor. Comment on the result.</p>	Volume %	TBP (°C)	0	216	10	243	30	268	50	284	70	304	90	318	95	327	100	334	10M	CO3
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Q 3	A naphtha fraction has the following ASTM D86 distillation data: <table border="1" data-bbox="201 1484 665 1824"><thead><tr><th>Volume %</th><th>Temperature (°C)</th></tr></thead><tbody><tr><td>0.0</td><td>138.8</td></tr><tr><td>10.0</td><td>149.6</td></tr><tr><td>30.0</td><td>158.8</td></tr><tr><td>50.0</td><td>165.8</td></tr><tr><td>70.0</td><td>169.9</td></tr><tr><td>90.0</td><td>178.1</td></tr><tr><td>95.0</td><td>180.4</td></tr></tbody></table> <p>Obtain the TBP curve using Daubert method.</p>	Volume %	Temperature (°C)	0.0	138.8	10.0	149.6	30.0	158.8	50.0	165.8	70.0	169.9	90.0	178.1	95.0	180.4	10M	CO3		
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	Index number $i$	$A_i$	$B_i$		
	1	7.4012	0.6024		
	2	4.9004	0.7164		
	3	3.0305	0.8008		
	4	0.8718	1.0258		
	5	2.5282	0.8200		
	6	3.0419	0.7750		
	7	0.1180	1.6606		
Q 4	Discuss the Isomerization of Light naphtha in detail?			<b>10M</b>	<b>CO4</b>
Q 5	Explain Visbreaking process with reference to the following points only with a suitable diagram. a) Feed stock b)Visbreaking Reactions c)Visbreaking Severity d) Kinetics of Visbreaking e) Product Yield and Properties			<b>10M</b>	<b>CO4</b>
Q 6	a)Find the catalyst volume needed for the desulphurization of VGO. The initial Sulphur content is 2.3 wt% and the final Sulphur content of the product is 0.1 wt%. The reaction rate constant ( $h^{-1}$ ) can be expressed as : $k = 2.47 \times 10^{10} \exp\left(\frac{-14,995}{T}\right)$ The reaction conditions are $T=415^{\circ}C$ and $P=5.1$ M Pa. The order of the reaction was found to be $n=1.7$ . The feed flow rate is 167,500 kg/h and has a density of 910 kg/ $m^3$ . b)What is the role of Hydrotreating in a refinery?			<b>10M</b>	<b>CO5</b>
<b>SECTION B (Attempt any TWO questions)</b>					
Q 7	a)Explain how vacuum is maintained in the vacuum distillation unit. <b>10M</b> b) Write a note on the auxiliary equipment used in ADU and VDU? <b>10M</b>			<b>20 M</b>	<b>CO3</b>
Q 8	a) With the help of a neat-labeled diagram, explain the working of a fluid catalytic cracking (FCC) unit. <b>10M</b> b) What are the chemical reactions involved in the catalytic alkylation and also write a note on the catalysts used? <b>10M</b>			<b>20 M</b>	<b>CO4</b>
Q 9	a)With the help neat process flow diagram, describe the propane deasphalting process? <b>10M</b> b)Discuss the method of phenol extraction used in treating of lubes. <b>10M</b>			<b>20 M</b>	<b>CO5</b>