


Name:			
Enrolment No:			
<p style="text-align: center;"><b>UPES</b> <b>End Semester Examination, May 2025</b></p>			
<b>Course: Medicinal Chemistry-I</b> <b>Program: B. Pharm</b> <b>Course Code: BP402T</b>		<b>Semester: IV</b> <b>Duration: 03 Hours</b> <b>Max. Marks: 75</b>	
<b>Instructions:</b> No additional material like graph paper, log table, <i>etc</i> is allowed for this examination.			
<b>SECTION A</b> <b>(20 Q x 1 M = 20 Marks)</b>			
<b>S. No.</b>	<b>Attempt all questions from section A.</b>	<b>Marks</b>	<b>COs</b>
<b>Q 1</b>	Which of the following is hydantoin analogs? a) Phensuccimide      b) Phenytoin c) Piroxicam            d) Zomepirac	<b>1</b>	<b>CO1</b>
<b>Q 2</b>	Most weakly basic drugs (pKa > 8) are absorbed from... a) Stomach              b) Intestine c) Both                  d) None of the above	<b>1</b>	<b>CO1</b>
<b>Q 3</b>	Generally, drugs are absorbed in which form? a) In ionized form      b) In unionized form c) In both of above form   d) In none of above form	<b>1</b>	<b>CO1</b>
<b>Q 4</b>	.....is the NSAIDs drug, which Indole derivative. a) Mefenamic acid      b) Ibuprofen c) Piroxicam              d) Indomethacin	<b>1</b>	<b>CO1</b>
<b>Q 5</b>	Choose the basic nucleus present in the sympathomimetic agents. a) Catechol nucleus    b) Benzyl nucleus c) Naphthol              d) Indole	<b>1</b>	<b>CO1</b>
<b>Q 6</b>	Draw the structure of Ibuprofen.	<b>1</b>	<b>CO1</b>
<b>Q 7</b>	.....is Ultra-short-acting Barbiturates. a) Phenobarbitone      b) Butobarbitone c) Pentobarbitone      d) Thiopentone	<b>1</b>	<b>CO1</b>
<b>Q 8</b>	Enlist Phase-I reactions.	<b>1</b>	<b>CO1</b>
<b>Q 9</b>	The most significant protein involved in binding with drug is..... a) Albumin      b) Glycoprotein c) Lipoprotein   d) Globulin	<b>1</b>	<b>CO1</b>
<b>Q 10</b>	Replacement of oxygen at C-2 position of barbituric acid by a sulfur atom ..... a) Has no change in the activity      b) Increases the activity c) Decreases the activity              d) Show anxiolytic activity	<b>1</b>	<b>CO1</b>
<b>Q 11</b>	Write the structure of Phenylephrine.	<b>1</b>	<b>CO2</b>
<b>Q 12</b>	Write the structure of carbachol.	<b>1</b>	<b>CO2</b>
<b>Q 13</b>	Draw the structure of Aspirin.	<b>1</b>	<b>CO2</b>
<b>Q 14</b>	Which type of ring system found in Diazepam?	<b>1</b>	<b>CO2</b>
<b>Q 15</b>	Propranolol is prepared by condensing..... a) $\alpha$ -naphthol and epichlorohydrin      b) $\alpha$ -naphthol and chloropropanol c) Phenol and epichlorohydrin          d) Chloro naphthol and propanol	<b>1</b>	<b>CO2</b>
<b>Q 16</b>	Define Bioisoster with an example.	<b>1</b>	<b>CO2</b>

<b>Q 17</b>	Carbachol differs from acetylcholine by..... a) Ester                                      b) Amide c) Chloro group                              d) Hydroxyl group	<b>1</b>	<b>CO2</b>
<b>Q 18</b>	Introduction of methyl group at alpha ( $\alpha$ ) position of acetylcholine forms acetyl- $\alpha$ -methyl choline which has more selectivity towards..... a) Nicotinic receptor                      b) Muscarinic receptor c) Both                                          d) None of the above	<b>1</b>	<b>CO2</b>
<b>Q 19</b>	Loxapine belongs to..... derivative a) Dihydroindole                          b) Phenothiazine c) Dibenzoxazepine                      d) Diphenylbutyl piperidines	<b>1</b>	<b>CO2</b>
<b>Q 20</b>	Barbituric acid is prepared by the condensation of..... a) Malonic acid and urea                      b) Diethylmalonate and urea c) Malonic acid with methyl urea              d) diethylmalonate with methyl urea	<b>1</b>	<b>CO2</b>
<b>SECTION B (20 Marks)</b> <b>(2 Q x 10 M = 20 Marks)</b>			
	<b>Attempt any two questions from section B.</b>	<b>Marks</b>	
<b>Q 1</b>	Classify the cholinergic receptors. Explain the breakdown of acetylcholine and the SAR of a direct acting parasympathomimetic drug.	<b>(2+8)</b>	<b>CO3</b>
<b>Q 2</b>	A. Give biosynthesis, metabolism of nor-adrenaline. B. Write the classification and SAR of adrenergic agents.	<b>(4+6)</b>	<b>CO3</b>
<b>Q 3</b>	A. Define the terms sedative and hypnotic. B. Classify and explain the SAR and mechanism of barbiturates.	<b>(2+8)</b>	<b>CO4</b>
<b>SECTION-C (35 Marks)</b> <b>(7 Q x 5 M = 35 Marks)</b>			
	<b>Attempt any seven questions from section C.</b>	<b>Marks</b>	
<b>Q 1</b>	Outline the synthetic step for Mefenamic acid	<b>5</b>	<b>CO3</b>
<b>Q 2</b>	Explain the catabolism of acetyl choline and explain the SAR of direct acting parasympathomimetic agent.	<b>(2+3)</b>	<b>CO3</b>
<b>Q 3</b>	Classify general anesthetics with examples. Outline the synthesis of Ketamine.	<b>(3+2)</b>	<b>CO3</b>
<b>Q 4</b>	Discuss following physicochemical properties of drug in relation to biological action. I) Isomerism    II) Partition coefficient	<b>(2.5+2.5)</b>	<b>CO4</b>
<b>Q 5</b>	Discuss SAR and Classification of Morphine Analogs.	<b>(2.5+2.5)</b>	<b>CO4</b>
<b>Q 6</b>	Give an account on inhalation anesthetics.	<b>5</b>	<b>CO4</b>
<b>Q 7</b>	Discuss in detail SAR of Benzodiazepines as sedative and hypnotics.	<b>5</b>	<b>CO4</b>
<b>Q 8</b>	What are hydantoins? Write the chemistry of hydantoins.	<b>5</b>	<b>CO5</b>
<b>Q 9</b>	Give an account on reversible and irreversible Cholinesterase inhibitors.	<b>5</b>	<b>CO5</b>