Name:

**Enrolment No:** 



## **UPES**

## **End Semester Examination, May 2025**

Course: Radiation Safety Semester: IV
Program: BSc Physics by research Time: 03 hrs.

Course Code: HSFS 2026K Max. Marks: 100

## **Instructions:**

1. All questions are mandatory in Section A.

2. Internal choices are given in questions #9 and #11 of Sections B and C respectively.

**3.** A list of important constants is provided at the end of this question paper.

## SECTION A (5Qx4M=20Marks)

S. No.		Marks	CO
Q 1	Compute the radiation absorbed dose in air corresponding to 1 Roentgen exposure. For air the average ionization energy is 34 J/ion.	4	CO2
Q 2	Illustrate the current voltage relationship for a gas filled detector, mentioning the regions to be used as counters.	4	CO3
Q 3	Describe the three radiation protection principles.	4	CO1
Q 4	Explain the concept of Annual Limit of Intake (ALI) and Derived Air Concentration (DAC) in the context of radiation protection.	4	CO1
Q 5	Explain the detection mechanism of neutrons.	4	CO2
	SECTION B		
	(4Qx10M= 40 Marks)		
Q 6	Briefly describe the principle, construction and working of proportional counter.	10	CO1
Q 7	Show that in Compton scattering the direction of the recoil electron is given as $tan\emptyset = \frac{cot\frac{\theta}{2}}{1+\frac{\hbar\nu}{m_0c^2}}$ . Where the symbols have their usual meaning.	10	CO3
Q 8	Discuss the internal and external exposure of radiation.	10	CO2
Q 8 Q 9	A patient receives an injection of $1.1 \times 10^8$ Bq of $^{131}$ I, which gets accumulated in the thyroid (mass of thyroid $\approx 2$ gm). Mean energy of the emitted radiation is $E \approx 300$ keV. Determine the dose rate to the thyroid.   OR  Determine the thickness of a lead shield required to reduce the exposure rate due to Cs-137 from 100 R/hr to 1R/hr. mass attenuation coefficient for lead is $0.11 \text{ cm}^2/\text{gm}$ and density of lead is $11.3 \text{ gm/cm}^3$ .	10	CO2

Q 10  a) Describe the various interaction mechanism of gamma rays with matter.  b) A gamma ray photon interacts with electron and gets scattered at an angle $\theta$ . Obtain expression for wavelength shift.  10  Q 11  a) Explain the principle, construction and working of multi wire proportional counter  b) Illustrate the functioning of projection imaging gamma camera.  OR  a) Describe briefly the Accelerator driven subcritical system for nuclear waste transmutation.  b) Illustrate the nuclear technique for industrial applications such as nuclear gauges and material modification.  Planck's constant, $h = 6.6 \times 10^{-34} \text{ J}$ ,s  Boltzmann's constant, $k = 1.38 \times 10^{-23} \text{ J/K}$ Mass of electron, $m_e = 9.1 \times 10^{-31} \text{ Kg}$ Charge on an electron, $e = 1.67 \times 10^{-27} \text{ Kg}$ Speed of light, $c = 3 \times 10^{6} \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^{7} \text{ m-1}$ Avogadro's number $= 6.023 \times 10^{23}$ Permeability of free space, $\mu_o = 4\pi \times 10^{-7} \text{ Henry/m}$		(2Qx20M=40 Marks)		T-
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	Q 10		10	
Q 11  a) Explain the principle, construction and working of multi wire proportional counter  b) Illustrate the functioning of projection imaging gamma camera.  OR  CO3  a) Describe briefly the Accelerator driven subcritical system for nuclear waste transmutation.  b) Illustrate the nuclear technique for industrial applications such as nuclear gauges and material modification.  Planck's constant, $h = 6.6 \times 10^{-34} \text{ J.s}$ Boltzmann's constant, $k = 1.38 \times 10^{-23} \text{ J/K}$ Mass of electron, $m_e = 9.1 \times 10^{-31} \text{ Kg}$ Charge on an electron, $e = 9.1 \times 10^{-31} \text{ Kg}$ Charge on an electron, $e = 1.6 \times 10^{-19} \text{ C}$ Mass of proton, $m_p = 1.67 \times 10^{-27} \text{ Kg}$ Speed of light, $c = 3 \times 10^{8} \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^{7} \text{ m-1}$ Avogadro's number $= 6.023 \times 10^{23}$				CO3
proportional counter b) Illustrate the functioning of projection imaging gamma camera.  OR  OR  a) Describe briefly the Accelerator driven subcritical system for nuclear waste transmutation. b) Illustrate the nuclear technique for industrial applications such as nuclear gauges and material modification.  Planck's constant, $h = 6.6 \times 10^{-34} \text{ J.s}$ Boltzmann's constant, $k = 1.38 \times 10^{-23} \text{ J/K}$ Mass of electron, $m_e = 9.1 \times 10^{-31} \text{ Kg}$ Charge on an electron, $e = 1.6 \times 10^{-19} \text{ C}$ Mass of proton, $m_p = 1.67 \times 10^{-27} \text{ Kg}$ Speed of light, $c = 3 \times 10^8 \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^7 \text{ m-1}$ Avogadro's number $= 6.023 \times 10^{23}$		an angle $\theta$ . Obtain expression for wavelength shift.	10	
b) Illustrate the functioning of projection imaging gamma camera.  OR  a) Describe briefly the Accelerator driven subcritical system for nuclear waste transmutation. b) Illustrate the nuclear technique for industrial applications such as nuclear gauges and material modification.  Planck's constant, $h = 6.6 \times 10^{-34} \text{ J.s.}$ Boltzmann's constant, $k = 1.38 \times 10^{-23} \text{ J/K}$ Mass of electron, $m_e = 9.1 \times 10^{-31} \text{ Kg}$ Charge on an electron, $e = 1.6 \times 10^{-19} \text{ C}$ Mass of proton, $m_p = 1.67 \times 10^{-27} \text{ Kg}$ Speed of light, $c = 3 \times 10^8 \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^7 \text{ m-1}$ Avogadro's number $= 6.023 \times 10^{23}$	Q 11		10	
a) Describe briefly the Accelerator driven subcritical system for nuclear waste transmutation. b) Illustrate the nuclear technique for industrial applications such as nuclear gauges and material modification.  Planck's constant, $h = 6.6 \times 10^{-34} \text{ J.s}$ Boltzmann's constant, $k = 1.38 \times 10^{-23} \text{ J/K}$ Mass of electron, $m_e = 9.1 \times 10^{-31} \text{ Kg}$ Charge on an electron, $e = 1.6 \times 10^{-19} \text{ C}$ Mass of proton, $m_p = 1.67 \times 10^{-27} \text{ Kg}$ Speed of light, $c = 3 \times 10^8 \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^7 \text{ m-1}$ Avogadro's number $= 6.023 \times 10^{23}$		1 1	10	
a) Describe briefly the Accelerator driven subcritical system for nuclear waste transmutation. b) Illustrate the nuclear technique for industrial applications such as nuclear gauges and material modification.  Planck's constant, $h = 6.6 \times 10^{-34} \text{ J.s}$ Boltzmann's constant, $k = 1.38 \times 10^{-23} \text{ J/K}$ Mass of electron, $m_e = 9.1 \times 10^{-31} \text{ Kg}$ Charge on an electron, $e = 1.6 \times 10^{-19} \text{ C}$ Mass of proton, $m_p = 1.67 \times 10^{-27} \text{ Kg}$ Speed of light, $c = 3 \times 10^8 \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^7 \text{ m-1}$ Avogadro's number $= 6.023 \times 10^{23}$		OR		CO2
nuclear waste transmutation. b) Illustrate the nuclear technique for industrial applications such as nuclear gauges and material modification.  Planck's constant, $h = 6.6 \times 10^{-34} \text{ J.s}$ Boltzmann's constant, $k = 1.38 \times 10^{-23} \text{ J/K}$ Mass of electron, $m_e = 9.1 \times 10^{-31} \text{ Kg}$ Charge on an electron, $e = 1.6 \times 10^{-19} \text{ C}$ Mass of proton, $m_p = 1.67 \times 10^{-27} \text{ Kg}$ Speed of light, $c = 3 \times 10^8 \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^7 \text{ m-1}$ Avogadro's number $= 6.023 \times 10^{23}$		a) Describe briefly the Accelerator driven subcritical system for		COS
Planck's constant, $h = 6.6 \times 10^{-34} \text{ J.s}$ Boltzmann's constant, $k = 1.38 \times 10^{-23} \text{ J/K}$ Mass of electron, $m_e = 9.1 \times 10^{-31} \text{ Kg}$ Charge on an electron, $e = 1.6 \times 10^{-19} \text{ C}$ Mass of proton, $m_p = 1.67 \times 10^{-27} \text{ Kg}$ Speed of light, $c = 3 \times 10^8 \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^7 \text{ m-1}$ Avogadro's number $= 6.023 \times 10^{23}$		· · · · · · · · · · · · · · · · · · ·	10	
Planck's constant, $h = 6.6 \times 10^{-34} \text{ J.s}$ Boltzmann's constant, $k = 1.38 \times 10^{-23} \text{ J/K}$ Mass of electron, $m_e = 9.1 \times 10^{-31} \text{ Kg}$ Charge on an electron, $e = 1.6 \times 10^{-19} \text{ C}$ Mass of proton, $m_p = 1.67 \times 10^{-27} \text{ Kg}$ Speed of light, $c = 3 \times 10^8 \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^7 \text{ m-1}$ Avogadro's number $= 6.023 \times 10^{23}$		b) Illustrate the nuclear technique for industrial applications such		
Boltzmann's constant, $k = 1.38 \times 10^{-23}$ J/K Mass of electron, $m_e = 9.1 \times 10^{-31}$ Kg Charge on an electron, $e = 1.6 \times 10^{-19}$ C Mass of proton, $m_p = 1.67 \times 10^{-27}$ Kg Speed of light, $c = 3 \times 10^8$ m/s Rydberg Constant, $R = 1.097 \times 10^7$ m-1 Avogadro's number = $6.023 \times 10^{23}$		as nuclear gauges and material modification.	10	
Mass of electron, $m_e = 9.1 \times 10^{-31} \text{ Kg}$ Charge on an electron, $e = 1.6 \times 10^{-19} \text{ C}$ Mass of proton, $m_p = 1.67 \times 10^{-27} \text{ Kg}$ Speed of light, $c = 3 \times 10^8 \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^7 \text{ m-1}$ Avogadro's number = $6.023 \times 10^{23}$				-
Charge on an electron, $e = 1.6 \times 10^{-19} \text{ C}$ Mass of proton, $m_p = 1.67 \times 10^{-27} \text{ Kg}$ Speed of light, $c = 3 \times 10^8 \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^7 \text{ m-1}$ Avogadro's number = $6.023 \times 10^{23}$				
Mass of proton, $m_p = 1.67 \times 10^{-27} \text{ Kg}$ Speed of light, $c = 3 \times 10^8 \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^7 \text{ m-1}$ Avogadro's number = $6.023 \times 10^{23}$				
Speed of light, $c = 3 \times 10^8 \text{ m/s}$ Rydberg Constant, $R = 1.097 \times 10^7 \text{ m-1}$ Avogadro's number = $6.023 \times 10^{23}$				
Rydberg Constant, $R = 1.097 \times 10^7 \text{ m-1}$ Avogadro's number = $6.023 \times 10^{23}$				
Avogadro's number = $6.023 \times 10^{23}$				