Name:	<b>WUPES</b>
<b>Enrolment No:</b>	UNIVERSITY OF TOMORROW

## **UPES**

## **End Semester Examination, May 2025**

Programme Name: B. Sc Geology Semester : II
Course Name : Igneous Petrology Time : 03 hrs
Course Code : PEGS1010 Max. Marks: 100

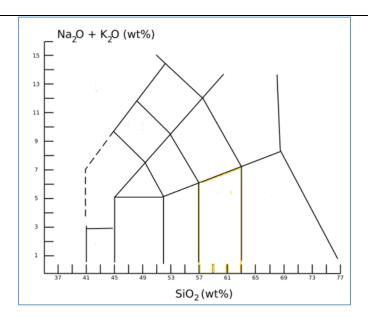
Nos. of page(s): 04

S.

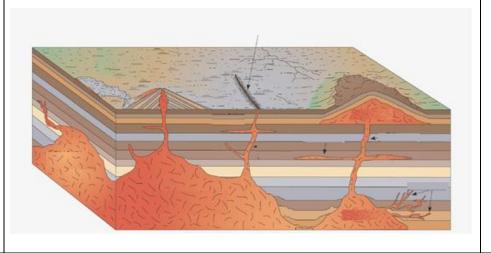
Attach the Binary Phase diagram with Answer-script Attach the CIPW sheet with Answer script Draw the diagram for Q 8 in Answer script

## **SECTION A (20 Marks)**

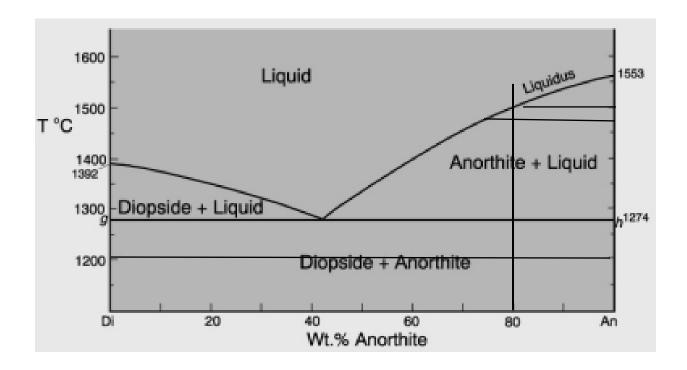
S. No.		Marks	CO									
Q 1	Classify igneous rocks using tabular classification scheme	05	CO1									
Q 2	Explain the megascopic textures of Igneous rock	05	CO1									
Q 3	05	CO2										
Q 4	Differentiate between Cone-sheet and ring-dyke	05	CO2									
SECTION B (40 Marks)												
Q 5	Define the following	02*5 =10	CO1									
Q 6	Explain Ophitic and poikilitic texture, highlighting the relation between the two.	10	CO2									
Q7	<ul><li>a) Summarize degree of crystallinity and classify igneous texture based on the same</li><li>b) ii. Classify ophitic texture in Igneous rocks</li></ul>	5+5 =10	CO2									
Q 8	Using the TAS classification  a) Classify the Volcanic igneous rocks  b) Explain Phonolite  c) Differentiate between Basalt and Picro-basalt	10	CO3									



**OR**Identify and label (marked ones and unmarked ones) the various igneous forms and describe their characteristic features.



	SECTION C (40 Marks)							
Q 9	Classify Basalt and analyze the role of garnet in formation of various types of basalt/ basaltic magma.	20	CO3					
Q 10	Using a binary isobaric diagram, calculate the percentage of solid and liquid, along with their respective composition at different temperature conditions. The temperature conditions are 1490°C (T1), 1400°C (T2), 1274°C and composition of magma: An80Di20.  OR  Using CIPW Norm, calculate the Salic and Femic minerals, their abundance, and the rock class.	20	CO3					



Consti	ituents Of Rock	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	Fe <sub>2</sub> O <sub>3</sub>	FeQ	MgO	CaQ	Na <sub>2</sub> O	K <sub>2</sub> O	H <sub>2</sub> O	CO <sub>2</sub>	TiO <sub>2</sub>	P2O5	SO <sub>2</sub>	S	MnO					
Percen	tages(analysis)	49.68	36.13	2.49	8.88	1.13	0.79	0.25	0.32						П		Molecular			Gro	up of
Molec	cular Weights	60	102	160	72	40	56	62	94	18	44	80	32	355	19	71	Proportions	Molecular	Percentage		dard
Molecular Proportion															П		Troporation	Weights 60	NORM		eral
Quartz	SiO <sub>2</sub>																			Q	
Orthoclase	K <sub>2</sub> O, Al <sub>2</sub> O <sub>3</sub> , 6SiO <sub>2</sub>																	556			
Albite	Na <sub>2</sub> O, Al <sub>2</sub> O <sub>3</sub> , 6 SiO <sub>2</sub>																	524			
Anorthite	CaO, Al <sub>2</sub> O <sub>1</sub> , 2 SiO <sub>2</sub>																	278		F	
Leucite	K2O, Al <sub>2</sub> O <sub>3</sub> , 4 SiO <sub>2</sub>																	436			
Nepheline	Na2O, Al <sub>2</sub> O <sub>3</sub> , 2 SiO <sub>2</sub>																	284		L	Salic
Corundum	Al <sub>2</sub> O <sub>3</sub>														П			102		С	Group
Acmite	Na <sub>2</sub> O, Fe <sub>2</sub> O3, 4SiO <sub>2</sub>																	462			
	CaO, SiO2																	116			
	MgO, SiO2																	100			
Diopside	FeO, SiO2																	132			
Wollastonite	CaO, SiO2														П			116			
	MgO, SiO2																	100			
Hypersthene	FeO, SiO2																	132		P	
	2MgO, SiO2																	140			
Olivine	2FeO, SiO2														П			204		0	
Magnetite	FeO, Fe <sub>2</sub> O <sub>3</sub>																	232			]
Haematite	Fe <sub>2</sub> O <sub>3</sub>																	160			
Ilmanite	FeO, TiO2																	152		M	
Pyrite	FeS <sub>2</sub>														П			120			
Apatite	3Cao, P <sub>2</sub> O <sub>3</sub> , 1/3CaF <sub>2</sub>																	336			Femic
Calcite	CaO.CO <sub>2</sub>																	100		A	Group