Name:

**Enrolment No:** 



## UPES End Semester Examination, May 2024

## Course: Chemistry of Advanced Organic Compounds Program: MSc Chemistry Course Code: CHEM7054

Semester: II Time : 03 hrs. Max. Marks: 100

Instructions: Read all the below mentioned instructions carefully and follow them strictly:

- 1) Mention Roll No. at the top of the question paper.
- 2) Do not write anything on the question paper except roll number.
- 3) Attempt all the parts of a question at one place only.
- 4) Internal choice is given only in Q 9 and 11.

## SECTION A (5Qx4M=20Marks)

S. No.		Marks	СО
Q 1	Complete the following reaction sequence:		
	$ \begin{array}{c} & \xrightarrow{\text{HNO}_3} \\ \hline & \xrightarrow{\text{Room}} \\ & \xrightarrow{\text{temp.}} \end{array} \end{array} \xrightarrow{\text{HI}} \mathbf{A} \xrightarrow{\text{[H]}} \mathbf{B} \xrightarrow{\text{NaNO}_3/\text{HCl}} \mathbf{C} \xrightarrow{\text{Cu}_3\text{Cl}_3/\text{HCl}} \mathbf{D} $	4	CO2
Q 2	Carry out following conversions:		
	a. Thiophene to sulfone.	4	CO2
	b. Thiophene to n-butane.	7	
Q 3	What happens when furfural reacts with:		
	a. Sodium hydroxide?	4	CO2
	b. Sodium acetate in the presence of acetic anhydride?		
Q 4	Name four reducing agents along with their formulae/ structures.	4	CO1
Q 5	Which of the two is stronger base: pyridine or piperidine? Justify your	4	CO1
	answer.		
	SECTION B		
	(4Qx10M= 40 Marks)		
Q 6	An organic compound 'A' (C10H8) on oxidation with acidic KMnO4		
	gives 'B' ( $C_8H_6O_4$ ), which on heating produces 'C' ( $C_8H_4O_3$ ). 'C' on		
	reaction with benzene in the presence of anhydrous AlCl <sub>3</sub> followed by	10	CO3
	$H_2SO_4$ gives 'D' (C <sub>14</sub> $H_8O_2$ ) which produces a compound containing		
	sulfur 'E' (C <sub>14</sub> H <sub>8</sub> O <sub>5</sub> S) on treatment with sulfuric acid. 'E' reacts with		

	NaOH to give sodium derivative of a compound 'F' ( $C_{14}H_8O_4$ ) which is a dye and converts to a trihydroxy compound 'G' ( $C_{14}H_8O_5$ ) on reaction with MnO <sub>2</sub> and H <sub>2</sub> SO <sub>4</sub> . Identify compounds 'A' to 'G', and complete the reaction sequence.		
Q 7	What happens when bromopyridine reacts with ammonia at $177 - 197$ °C? Which reaction is this? Also write the mechanism of this reaction.	10	CO3
Q 8	Cause the nitration of quinoline along with its mechanism.	10	CO3
Q 9	Why does isoquinoline give nucleophilic substitution reactions? Justify the suitable position for the attack of nucleophile in isoquinoline with an example. <b>OR</b> Prove that nitrogen atom in isoquinoline is present at 2 <sup>nd</sup> position in its structure.	10	CO2
	SECTION-C (2Qx20M=40 Marks)		
Q 10	<ul> <li>Write short notes on the following:</li> <li>a. NaBH<sub>4</sub></li> <li>b. Sodium cyanoborohydride</li> <li>c. Osmium tetroxide</li> <li>d. Aluminium isopropoxide</li> </ul>	20	CO1
Q 11	<ul> <li>a. Write the synthesis of indole starting from o-nitrophenol acetaldehyde. What is the basic process involved in this reaction?</li> <li>b. What happens when acetonyl acetone is heated with ammonia?</li> <li>c. Draw the structure of indigotin. Why is it named so? How can it be synthesized from naphthalene?</li> <li>OR</li> <li>a. How does indole react with chloroform in alkaline medium? Which reaction is this? Which type of intermediate is formed in this reaction?</li> <li>b. Suggest two chemical reactions which can prove that pyrrole behaves similar to phenol.</li> <li>c. Draw the structure of isatin along with its isomeric forms. Complete its synthesis starting from o-nitrobenzoyl chloride.</li> </ul>	5+5+10	CO1, CO2, CO2