Name:				10					
Enrolment No:									
UPES									
End Semester Examination, May 2024									
Course	n. B. Pharm		Semester: 1v Duration: 03 Hou	Irc					
Frogram: B. Fnarm     Duration: 03 Hot       Course Code: RP401T     May Marke: 75				115					
Instruct	tions: All the sections are compulsory.								
SECTION A									
1. Each	Ouestion will carry 1 Marks.								
2. Instr	uction: Select the correct answer(s)/ Objective	type questions.							
Answ	ers all the 20 questions.	type questions.							
S No	Ouestic	nc		Marks	COs				
0.1	Define Atronisomers	/115		1					
ŲI	Define Autopisoiners.			1	COI				
Q 2	Synonym of quinoline is.			1	CO2				
	(a) Benz-imidazole								
	(b)Benz-pyridine								
	(c) Benz-pyrrole								
	(d) Benz-oxazole								
03	Identify the position of nitrogen in isoquinoline	e ring.		1	CO1				
χv	(a) $1^{\text{st}}$	, ing.		-	001				
	(b) 2 <sup>nd</sup>								
	(c) $1^{st}$ and $2^{nd}$ both								
	(d) None of the above								
04	Identify how many nitrogen atoms are present i	n nyrazole		1	CO1				
۷Ŧ	(a) 1	in pyrazoie.		1	COI				
	(b) 2								
	(c) 3								
	(d) 4								
0.5	Which of the following is a six membered beta	roqualia ring?		1	CO1				
<b>V</b> 3	(a) Pyrrole			1	COI				
	(b) Pyridine								
	(c) Furan								
	(d) Pyrazole								
0(	Which magant is used in Deiman Times	tion of press		1	<u> </u>				
۷o	(a) Dichloromothano	tion of pyrrole?		1	02				
	(a) Diemoromethane (b) Chloroform								
	(c) Carbon dioxide								
	(d) Formic acid								
Q 7	Which statement is true for azepine ring?			1	CO2				
	(a) Six membered ring with one nitrogen (b) Six membered ring with one nitrogen and a	ne ovugen							
	(b) Six membered ring with one nitro set	ne oxygell							
	(c) Seven membered ring with one nitrogen								
	(a) Seven membered ring with one oxygen								

Q 8	Identify the product when 1,4-diketone reacts with phosphorus trisulphide. (a) Pyrrole (b) Furan (c) Thiophene		CO2
	(d) Pyridine		
Q 9	Cyclic compounds can exhibit geometrical isomerism: True or False.	1	CO1
Q 10	According to CIP Sequence rule, the complete sequence of priority is (a) $-OH > -CH2OH > -CHO > -H$ (b) $-H > -CH2OH > -CHO > -OH$ (c) $-H > -OH > -CH2OH > -CHO$ (d) $-OH > CHO > -CH2OH > -H$	1	CO2
Q 11	Define constitutional isomers.	1	CO1
Q 12	<ul> <li>Which reaction proceeds in the presence of hydrazoic acid?</li> <li>(a) Dakin reaction</li> <li>(b) Beckmann rearrangement</li> <li>(c) Wolff Kishner reaction</li> <li>(d) Schmidt reaction</li> </ul>	1	CO2
Q 13	Choose catalyst used in Oppenauer oxidation. (a) Hydrogen peroxide (b) Aluminium isopropoxide (c) Aluminium chloride (d) Hydrazine hydrate	1	CO2
Q 14	Define enantiomers.	1	CO1
Q 15	Identify the product when benzaldehyde reacts with hydrazine hydrate in presence of ethanol. (a) Acetophenone (b) Toluene (c) Cresol (d) Ethyl benzene	1	CO2
Q 16	<ul> <li>Which reaction is used for the synthesis of lactam from cyclic oximes?</li> <li>(a) Dakin reaction</li> <li>(b) Beckmann rearrangement</li> <li>(c) Wolff Kishner reaction</li> <li>(d) Schmidt reaction</li> </ul>	1	CO1
Q 17	Define stereoisomerism.	1	CO1
Q 18	Assign R or S configuration for following: OH H <sub>2</sub> N H <sub>2</sub> OH	1	CO1
Q 19	What do you understand by the term center of symmetry?	1	CO2

Q 20	Enantiomers have same boiling point: True or False.	1	CO1				
SECTION B (20 Marks)							
(2Qx10M=20 Marks)							
1. Each	question will carry 10 marks.						
2. Instr	uction: Long Answer type questions (Answer any two questions out of three questions)						
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Q 1	1 Define conformational isomers and explain with suitable example how they differ from		CO3				
	constitutional isomers. Discuss the concept of comormational stability in n-outaile.						
Q 2	Discuss the methods of resolution of racemic modification in detail. Highlight the	(6+4)	CO2				
	advantages and limitations of each method.						
0.2	(a) Evaluin electrophilic substitution reactions of this have with suitable examples	(5,5)	COS				
Ų3	(a) Explain electrophilic substitution reactions of thiophene with suitable examples. (b) Identify the product and write appropriate reaction when pyrrole undergoes:		COS				
	(i) Reduction (ii) Oxidation.						
	SECTION C (25 Marks)						
	(7Qx5M=35 Marks)						
1. Each 2. Instr	question will carry 5 marks. uction: Short Answer type questions						
Answ	er any seven questions out of nine questions.						
01	Describe the aromaticity and resonance in pyrrole with suitable resonating structures in	5	CO1				
Q I	detail.	5	COI				
		(2.5.	005				
Q 2	<b>Q</b> 2 Mention the product and write suitable reactions if acridine reacts with (a) nitrating mixture (b) sodamide in presence of liquid ammonia.		COS				
Q 3	Explain the structure and uses (in pharmaceuticals) of indole and azepine.		CO1				
Q 4	Explain Skaup synthesis and Doebner miller synthesis for preparation of quinoline		CO5				
0.	derivatives with suitable examples.		<u> </u>				
Q5	What will be the product of reaction of acetone with benzaldehyde in the presence of aqueous sodium hydroxide? Justify the product formation with appropriate mechanism.		CO4				
Q 6	Write down reaction with appropriate mechanism related to reduction of carbonyl compounds by LiAlH. What precautions should be taken while handling LAH?		CO4				
Q 7	Differentiate Clemmenson and Wolff-Kishner reduction with suitable examples.	5	CO4				
Q 8	<b>Q 8</b> Justify with suitable example:		CO3				
	a) Meso compounds can be optically active: true or false.	2.5)					
	b) A compound with chiral center cannot be optically inactive" true or false. Write a short note on D & L and Syn & Anti system of nomenclature of optical isomers	5	$CO^{2}$				